



## Formation of 1,3,4-oxadiazolines and 1,3,4-oxadiazepines through acetylation of salicylic hydrazones



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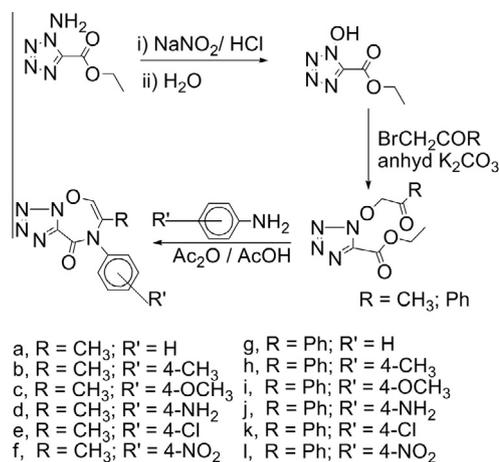
### ABSTRACT

A new series of 1,3,4-oxadiazolines and 1,3,4-oxadiazepines are prepared in a one-step reaction through cyclization of various *N*-benzylidene-2-hydroxybenzohydrazides. Cyclization in acetic anhydride yielded 1,3,4-oxadiazolines, while the reaction carried out in acetic anhydride–acetic acid gave 1,3,4-oxadiazepines, in some cases.

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Oxadiazolines and oxadiazepines are important compounds for both chemical and biological purposes.<sup>1,2</sup> They have been used extensively as synthons in various organic syntheses such as for the preparation of spiro-fused  $\beta$ -lactam oxadiazolines<sup>3</sup> and of fused oxadiazepines used as gamma secretase modulators for the treatment of Alzheimer's disease.<sup>4</sup> In addition, oxadiazolines and oxadiazepines have been reported to exhibit diverse pharmacological properties,<sup>5</sup> which include antimicrobial,<sup>6</sup> cytotoxic,<sup>7</sup> antifungal, and anticancer activities.<sup>8</sup> Various aldehyde and ketone acyl hydrazones have been cyclized to give 3-acyl-1,3,4-oxadiazolines under acylating conditions.<sup>9,10</sup> However, there are only three reports on acylhydrazones with a hydroxyl group at the *ortho* position of the benzene ring being cyclized to give 3-acyl-1,3,4-oxadiazolines.<sup>11</sup> In the case of oxadiazepines, several methods have been reported for their synthesis, all of which are multi-step in nature.<sup>6,12–14</sup> For example, El Badry and Taha<sup>2</sup> reported that the diazotization of ethyl 1-aminotetrazole-5-carboxylate in the presence of water resulted in the formation of ethyl 1-hydroxytetrazole-5-carboxylate. (Scheme 1).

Condensation of ethyl 1-hydroxytetrazole-5-carboxylate with bromoacetone and/or phenacyl bromide in absolute ethanol in the presence of anhydrous potassium carbonate provided acetyl-oxo and 2-oxyacetophenone compounds, which were then reacted

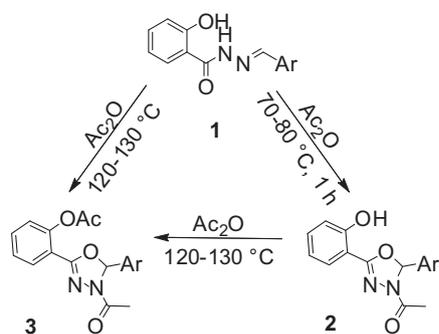


Scheme 1. El Badry and Taha's work.<sup>2</sup>

with various 4-substituted anilines in the presence of acetic anhydride/acetic acid to give 7-methyl(phenyl)-8-aryltetrazolo[1,5-*b*]-1,2,5-oxadiazepin-9-ones in three steps.<sup>6</sup> Herein, we report a novel, one-step intramolecular oxidative cyclization of a variety of substituted benzaldehyde acylhydrazones **1**<sup>15</sup> with a free hydroxyl group at the *ortho* position to give the oxadiazolines **2**, which

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**Scheme 2.** Synthesis of 1,3,4-oxadiazoline derivatives **3**.

were isolated and characterized. Acetylation of the oxadiazoline **2** led to the formation of 1,3,4-oxadiazoline derivatives **3**<sup>16–18</sup> as shown in **Scheme 2**. Thus, reactions of substituted benzaldehyde acylhydrazones **1** in acetic anhydride at 120–130 °C resulted in the cyclized products **3** (**Scheme 2**). The reactions proceeded smoothly with no side products being observed under these conditions.<sup>15,19</sup>

Under these acylation conditions, compounds **1a–i**, possessing either electron-donating or electron-withdrawing substituents on the aryl ring cyclized to give 1,3,4-oxadiazolines **3a–i**<sup>16–18,20</sup> in 58–85% yields<sup>18</sup> (**Table 1**). The presence of an electron-withdrawing substituent on the phenyl ring tended to give better yields with the best yield being obtained with a nitro substituent, and the lowest with a *tert*-butyl substituent.<sup>17,18</sup> This is to be expected since a strong electron-withdrawing group such as NO<sub>2</sub> on the aryl ring would enhance the electrophilicity of the iminium carbon, while an electron-donating group would decrease the electrophilicity.<sup>18</sup>

In some cases, when the cyclization reactions of **1** were carried out at 50–60 °C in acetic anhydride/acetic acid solution, 1,3,4-oxadiazepines **4** were obtained instead of 1,3,4-oxadiazolines **3** (**Scheme 3**).<sup>18,20,21</sup>

**Table 2** summarizes the products of the cyclization reactions of compound **1** using the Ac<sub>2</sub>O–AcOH conditions. Presumably, the acidic conditions influenced the reaction to form the seven-membered oxadiazepines.<sup>18,20,21</sup>

We have proposed two pathways leading to the formation of oxadiazolines **3** (**Scheme 4**). One pathway involves acetylation of the free hydroxyl group on the benzene ring to form **5**, which then undergoes intramolecular oxidative cyclization to form **3** (Pathway A). An alternative pathway involves intramolecular oxidative cyclization of **1** to first produce **2a** and **2b**, followed by acylation of the phenol to form **3** (Pathway B).

However, since we isolated only the oxadiazolines **2a** and **2b** with a free *ortho* phenolic group and no product **5** from this reaction, we concluded that the cyclization occurred through pathway B. Compounds **2a** and **2b** (**Scheme 4**), then underwent acetylation to produce **3a** and **3b**.

It has been well established that compound **1** can undergo keto–enol tautomerisation as shown in **Scheme 5**.

We propose that the mechanism for the oxidative cyclization reactions leading to **2a** and **2b** involves attack of the enolic oxygen of the enol tautomer on the azomethine imine moiety as shown in **Scheme 6**.

In the case of the seven-membered oxadiazepines, we propose that the reaction occurs via nucleophilic attack of the phenolic oxygen on the iminium carbon as shown in **Scheme 7**. Here, the iminium carbon acts as a carbonyl analogue and participates in an intramolecular nucleophilic addition reaction<sup>19,22</sup> with the *ortho* phenolic group. Subsequently, the oxadiazepine underwent acetylation to give only the diacetylated product **4** (**Scheme 7**).

**Table 1**  
Structures and yields of synthesized compounds **3a–i**

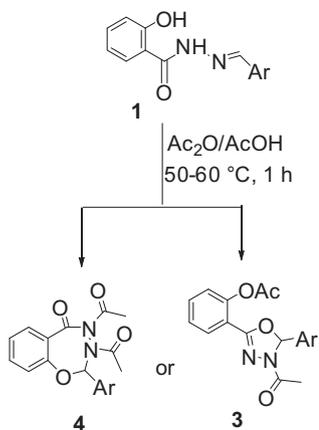
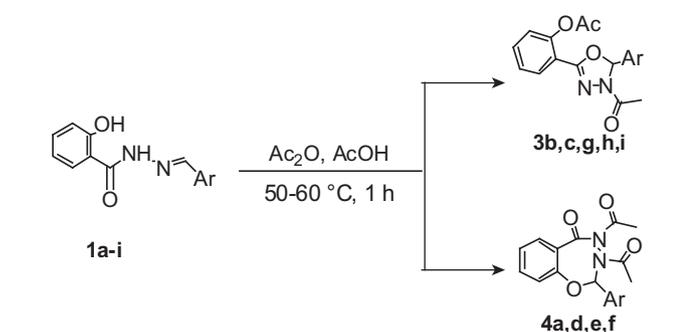
Entry	Ar	Product <sup>a</sup>	Yield <sup>c</sup> (%)
1			78
2			70
3			70
4			65
5			60
6			58
7			85
8			70
9			75

<sup>b</sup>Structure was confirmed by X-ray crystallography.

<sup>d</sup>This compound was previously reported in Ref. 11 along with a crystal structure, but without any data.

<sup>a</sup>All products were identified by ATIR, NMR, and EI-HRMS analyses.

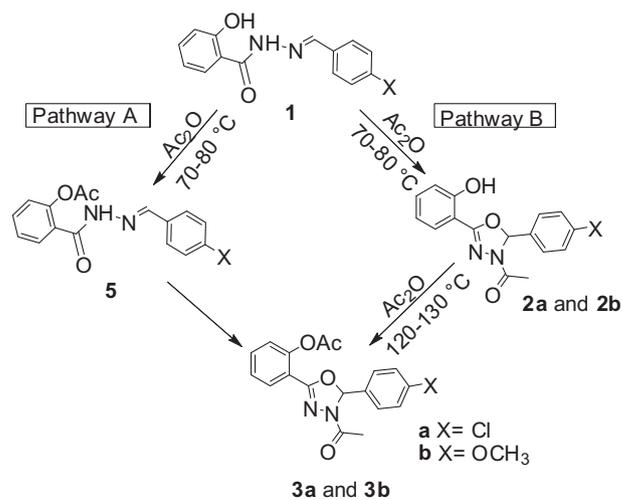
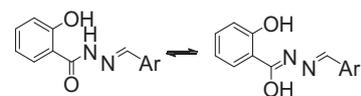
<sup>c</sup>Isolated yield after recrystallization.

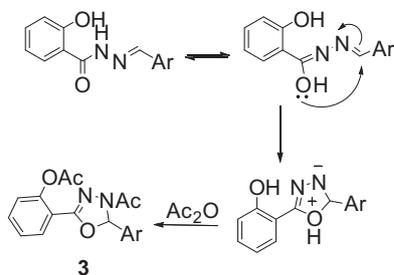
**Scheme 3.** Synthesis of 1,3,4-oxadiazepines **4** and 1,3,4-oxadiazolines **3**.**Table 2**  
Structures and yields of compounds **4a,d,e,f** and **3b,c,g,h,i**

Entry	Ar	Product <sup>a</sup>	Yield <sup>c</sup> (%)
1			63
2			75
3			66
4			55

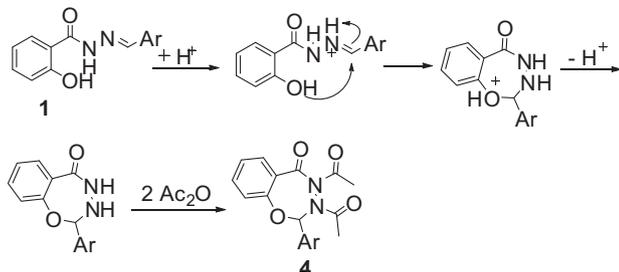
**Table 2 (continued)**

Entry	Ar	Product <sup>a</sup>	Yield <sup>c</sup> (%)
5			70
6			65
7			65
8			63
9			70

<sup>b</sup>Structure was confirmed by X-ray crystallography.<sup>a</sup>All products were identified by ATIR, NMR, and EI-HRMS analyses.<sup>c</sup>Isolated yield after recrystallization.**Scheme 4.** Proposed pathways for the cyclization of **1**.**Scheme 5.** Tautomerisation of compound **1**.



**Scheme 6.** A plausible mechanism for the formation of compounds **3a–i**.



**Scheme 7.** A plausible mechanism for the formation of compounds **4**.

In summary, 1,3,4-oxadiazolines containing an acetoxy group at the *ortho* position of the benzene ring were prepared in one-step, via intramolecular oxidative cyclization of acylhydrazones in acetic anhydride, which serves both as a reactant and the solvent. However, 1,3,4-oxadiazolines or 1,3,4-oxadiazepines were obtained in some cases, when the reactions were carried out under acid-catalyzed conditions.

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#### Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2014.12.037>.

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- General procedure for the synthesis of hydrazones **1a–i**: 2-Hydroxybenzohydrazide (0.30 g, 2 mmol) and different *para*-substituted benzaldehyde derivatives (0.2 g, 2 mmol) were refluxed in EtOH (20 ml) for 5 h. The solvent was removed by evaporation and the resulting products were obtained as white solids (**1a,c–i**) or as a yellow solid (**1b**).
- General procedure for synthesis of oxadiazoline analogs (Table 1): A mixture of hydrazone **1a–i** (1.58 mmol) in Ac<sub>2</sub>O (6 ml) was refluxed for 2 h under vigorous stirring. The solution was cooled and then poured onto crushed ice and stirred vigorously. A precipitate formed which was washed with distilled H<sub>2</sub>O to remove the Ac<sub>2</sub>O. The obtained solid was further purified by crystallization from an appropriate solvent.
- Most products were found to be homogeneous by TLC and 400 MHz NMR analyses, but when necessary, heterogeneous products were readily purified by silica gel column chromatography using hexane/CHCl<sub>3</sub> as the eluent.
- See Supplementary data for complete experimental details.
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- Crystallographic data for compounds **3c**, **3h**, **4a**, **4d** and **4e** have been deposited at the Cambridge Crystallographic Data Centre, with the deposition numbers 963045, 963044, 963041, 963043 and 963042, respectively.
- General procedure for the synthesis of oxadiazepine derivatives (Table 2): Compounds **4a,f,e,d** were obtained from the reaction of Ac<sub>2</sub>O (6 ml) with hydrazones **1a,f,e,d** (2 mmol) in the presence of AcOH (6 ml), and the resulting solution was stirred vigorously for 1 h at 50–60 °C. A precipitate formed which was washed with distilled H<sub>2</sub>O to remove the Ac<sub>2</sub>O. The obtained solid was further purified by crystallization from an appropriate solvent.
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